## Water Weakening and Fluid Rock Interactions in Chalks from the Mons Basin

Christian David Davide Geremia





# GEO<sup>2</sup>FRI<sup>2</sup>SK

GEOphysical and GEOtechnical impact

of Fluid-Rock Interactions

For RISK assessment in chalk formations

**Reservoir** applications

Geotechnical applications

#### **Reservoir applications**



Enhanced oil recovery operations

What is the impact of fluid substitution on the mechanical properties of a reservoir at depth?

#### Starting point of the project:

Design lab experiments mimicking oil-water substitution in reservoir rocks under stress



what happens in a reservoir at depth during the fluid substitution process ?



#### Oil-water substitution in reservoir rocks under stress

#### $\rightarrow$ Lab experiments on the Sherwood sandstone (UK)









at very low injection pressure < 1 MPa

#### Oil-water substitution in reservoir rocks under stress



Water injection can result in mechanical instability which can be monitored through seismic survey



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For RISK assessment in CHALK FORMATIONS Reservoir applications Geotechnical applications



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# Water-weakening and Fluid-Rock Interactions in Chalks from the Mons Basin

Presented by: Davide Geremia

Supervisors: Christian David, Beatriz Menéndez and Christophe Barnes







#### Outline

- 1. Water-Weakening: Underground geological reservoirs secondary recovery of oil
- 2. Water-Weakening: Comparison in between Obourg and Ciply chalk
- 3. Theories of water-weakening
- 4. Application: Underground cavities stability in abandoned quarries



#### Secondary Recovery







#### Secondary Recovery





Secondary Recovery





#### Secondary Rrecovery





Ekofisk Oil Field, North Sea



Phase 3: Chemical weakening and deformation



#### Experimental Approach - Secondary Recovery



#### Experimental Approach – Conventional Triaxial Tests



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#### Experimental Approach – Conventional Triaxial Tests



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#### Materials

**Composition:** ~ 100% Calcite (Voake et

al., 2019)

Grain density: 2.72 g/cm<sup>3</sup>

Bulk Density: 1.55 g/cm<sup>3</sup>

Mean Porosity: 43%

Permeability: 0.2 – 6 mD

Peak pore throat Radius (Mercury

**injection)** = 0.3 µm

Peak grain size (statistical):  $0.4 - 1.3 \ \mu m$ 

#### **Obourg chalk**



#### Ciply chalk



Composition: Calcite, Fluoroapatite Grain density: 2.73 g/cm<sup>3</sup> Bulk density: 1.68 g/cm<sup>3</sup> Mean Porosity: 39% Permeability: 40 mD





Materials

**Obourg chalk** 

Ciply chalk



Geremia et al. 2021a Geremia et al. 2021b



#### Results - Triaxial Tests







RockEnGeo.be

#### Planning Injection Tests





Axial Strain (%)

Geremia et al. 2021a



#### Methods - Injection Tests Vacuum









The Ciply chalk is more sensitive to water









## 2<sup>nd</sup> Part Comparing the two Chalks





#### Comparison Obourg and Ciply Chalk

Modified after Geremia et al. 2021b



#### Comparison Obourg and Ciply Chalk



Brazilian test: - Tensile strength

Triple point load test:

- Fracture toughness K<sub>IC</sub>
- Surface Energy





#### Geremia et al. 2021b

Through triaxial tests at low confining pressure:

- Cohesion
- Friction coefficient

### Comparison Obourg and Ciply Chalk – Wet to Dry Ratios

Geremia et al. 2021b











## 3<sup>rd</sup> Part The Mechanisms of Water-Weakening





Short-Term Mechanisms

#### **Repulsive Pressure**



#### **Surface Energy**

Energy to cut a solid body in two parts

Røyne et al. 2011 through double torsion experiment



#### Surface Energy and P\*

Carbonates

Sandstones



- Surface area
- Mineralogy



#### Surface Energy Estimation From Rachid Ismail's Internship







Surface Eenergy and Repulsive Pressure - AFM



Summary

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|          |                                       |          | Obourg Chal                 | k   |          | Ciply Chalk                 |  |   |
|----------|---------------------------------------|----------|-----------------------------|---|----------|-----------------------------|--|---|
| <b>→</b> | Surface Energy<br>(J/m <sup>2</sup> ) | Dry Rock | Water-<br>Saturated<br>Rock | $\lambda = \frac{\gamma^{(sat)}}{\gamma^{(dry)}}$ | Dry Rock | Water-<br>Saturated<br>Rock | $\lambda = rac{\gamma^{(sat)}}{\gamma^{(dry)}}$ |   |
| <b>→</b> | From contact<br>angle<br>measurements | 0.0234   | not<br>measurable           | х   | 0.0253   | not<br>measurable           | х  | Owens-Wendt model                                     |
|          | From AFM                              | 0.0207   | 0.0165                      | 0.80  | 0.0196   | 0.0141                      | 0.72   | $\gamma = \frac{F_{adh}}{4\pi R}$                     |
| <b>→</b> | From K <sub>ic</sub><br>measurements  | 0.5270   | 0.4150                      | 0.79  | 1.01     | 0.84                        | 0.83   | $\longrightarrow \qquad \gamma = \frac{K_{IC}^2}{2E}$ |

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$$P_{adh}^{(Dry,Wet)} = \frac{F_{adh}^{(Dry,Wet)}}{2\pi R^2}$$

*Repulsive pressure* =  $P_{adh}^{(Dry)} - P_{adh}^{(Wet)}$ 

|               | Obourg Chalk | Ciply Chalk    |
|---------------|--------------|----------------|
| Dry           | 5.06 MPa     | 4.79 MPa       |
| Wet           | 0.99 MPa     | 0.84 MPa       |
| Rep. Pressure | 4.1 MPa      | <u>3.9 MPa</u> |

 $C_{Dry} - C_{Wet} = 0.8 MPa$ 

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#### Surface Energy mechanism



Take home message:

• Results indicate that a hydration layer can decrease the surface energy, hence the energy to induce cracking

Repulsive Pressure mechanism



Geremia et al. 2021b

Take home message:

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- Results indicate that a hydration layer can also set up a repulsive pressure
- The presence of adsorbed ions dismantles the repulsive pressure









## 4<sup>th</sup> Part

## Influence of Cyclic Imbibition of Water on the Mechanical Properties of Ciply Chalk





#### *Ciclic Imbibition of Ciply Chalk – La Malogne Quarry*



Georgieva et al. 2020



*Ciclic Imbibition of Ciply Chalk – La Malogne Quarry* 



#### Experimental Approach – Imbibition/Drying Cycles

#### Ambient stress





#### Constant, not critical, axial load





*Ciclic Imbibition of Ciply Chalk – Ambient Stress* 



- 15 cycles of imbibition-evaporation with distilled water
- Imbibition: between 30 and 60 mins
- Evaporation: three days
- UCS and Young's modulus at cycle 0, 2, 6, 10, 15
- 5 samples for Young's modulus (load/unload cycle)
- 5 samples for UCS
- Environmental conditions:

Temperature range: 23-25 °C

Humidity range: 46 – 64 %

*Ciclic Imbibition of Ciply Chalk – Ambient Stress* 



Take home message:

- The mechanical behavior indicates heterogeneity in the rock samples
- UCS seems to be more affected by the porosity rather than cyclic imbibition UNIVERSITÉ RockEnGeo.be

#### *Ciclic Imbibition of Ciply Chalk – Ambient Stress*



Take home message:

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- Young's Modulus does not appear to be affected by cyclic imbibition
- Young's Modulus undergoes hardening

#### *Ciclic Imbibition of Ciply Chalk – Constant Load*

- Constant axial stress: 0.6 MPa
- 6 cycles of imbibition-evaporation with chemically equilibrated water
- Imbibition: around 60 mins
- Evaporation: two days





1° Imbibition – Constant Load





2° Imbibition – Constant Load











#### Take Home Message



Axial Strain (%)

1. Imbibition at ambient stress

Water saturation does not induce strain When dried, the rock strength is recovered

 Imbibition at constant not-critical load
Water saturation leads the rock sample from a dry state to a wet state producing irreversible strain
When drying, no strain is recovered
Hence, a new imbibition does not induce strain

3. Injection test at constant **critical** load

Being the constant applied stress much higher than its watersaturated strength, it fails catastrophically

- Knowing the stress state we can predict the resulting deformation/compaction
- Open question: What then causes higher damage in the transitional zone?
  Frost weathering?



#### General Conclusions

- Results indicate that a hydration layer can both decrease the surface energy and set up a repulsive pressure
- Changing the saturating fluid means changing completely the mechanical properties; the new properties can be quantified through conventional mechanical tests
- The mechanical strength reduces exponentially and progressively with the water saturation or wet volume

